Comparing the External Thermal Properties Between Lithium-Ion and Solid-State Electrolyte Batteries in Photovoltaic Applications

Problem Statement

As solar energy continues to rise in popularity, one key challenge remains: the role of batteries in ensuring its feasibility and reliability. Batteries store the energy generated by solar panels during peak sunlight hours and release it when the panels are operating at reduced efficiency, such as in the evening or at night.



A liquid-state electrolyte battery is the standard of what industries use for any battery-operated device. The cons of lithium-ion batteries are their flammability, unstableness, and subpar levels of efficiency at higher temperatures.



Solid-state electrolyte batteries are a better alternative due to enhanced user safety and higher ionic conductivity. This project aims to demonstrate why solid-state electrolyte batteries can revolutionize the solar industry.



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Ionic conductivity is important for batteries because it measures how well ions move through the battery's conductor. The semicircle shape shown above indicates good ionic conductivity.

LLZO Ramping Up at 373K (100C/212F)



This project combines tangible components presented through an in-person demonstration with experimental and simulation results showcased via a physical proof of concept and modeling simulation.





The physical proof of concept highlights the practical implementation of the project, demonstrating how solid-state electrolyte batteries could revolutionize the battery industry. This model can be scaled for various applications, including solar farms, data centers, residential buildings, bikes, and more. By adopting this technology, energy generation and distribution can become significantly more efficient and reliable.

At higher temperatures, the ionic conductivity increases, as shown by a larger semicircle. This happens because heat encourages more ion movement across the layers. Therefore, solid-state electrolyte samples perform well in high-temperature applications.

Proof of Concept

Conclusion

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